Ultrafast 25-fs relaxation in highly excited states of methyl azide mediated by strong nonadiabatic coupling

William K. Peters, David E. Couch, Benoit Mignolet, Xuetao Shi, Quynh L. Nguyen, Ryan C. Fortenberry, H. Bernhard Schlegel, Françoise Remacle, Henry C. Kapteyn, Margaret M. Murnane, and Wen Li

Highly excited neutral electronic states present many challenges and opportunities for exploring strongly correlated nonadiabatic dynamics. A distinguishing aspect of highly excited neutral molecular systems, compared with the well-studied lower excited states, is that these electronic states are closely spaced and strongly coupled, even in the Franck–Condon region, enabling new and faster relaxation pathways. Molecular dynamics at large internal energies impact a variety of natural phenomena, from thermosphere–mesosphere coupling to astrochemical processes (3, 4). However, comparatively little work has been done on these systems since they can only be safely handled in small quantities, and, until recently, no sources were available that could produce several-electronvolt tunable light with femtosecond time resolution to directly probe these states experimentally. Furthermore, highly excited electronic potential energy surfaces are challenging to describe theoretically—due to the large density of states and the fact that small structural changes can lead to large changes in the electronic character of the states.

When molecules are excited by vacuum-UV (VUV) light in the 7- to 10-eV range, just below the ionization potential, one or several excited states can be accessed depending on the bandwidth of the pulse. In this energy range, there is typically a dense manifold of valence and Rydberg states that are often strongly coupled, each with an associated vibrational structure (9). As a result, VUV molecular absorption spectra typically contain quasicontinuous but structured bands with overlapping sharp peaks, indicating the wide variety of timescales and states involved. This differs from the lowest few excited states, where dynamics are dominated by only one or two electronic states. Depending on the character of the electronic state or the band of excited states accessed by the VUV pulse, different relaxation pathways, such as nonadiabatic relaxation, fluorescence, internal conversion, or dissociation on the excited state may occur. Consequently, small changes in photon energy can lead to dramatically different energetic pathways that are still incompletely understood and are a focus of intense current research.

These relaxation pathways also have completely different timescales. For instance, molecules excited by visible light can relax either by fluorescence on nanosecond timescales or through conical intersections, typically on femtosecond to picosecond timescales. The relaxation through conical intersections—molecular conical intersection | nonadiabatic dynamics | Rydberg-valence electronic states | PEPICO

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Significance

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geometries where two or more electronic states come into a degeneracy—are known to play a ubiquitous role in the dynamics of electronically excited states (10, 11). In polyatomic molecules, conical intersections exist along multidimensional “seams,” and the population transfer can occur anywhere along the seam. The minimum energy along the seam is the minimum energy conical intersection (MECI) (12–14) and is the geometry usually reported in the literature. In the case of VUV excitation, it has long been known that a molecule will usually relax to its lowest excited state much faster than fluorescence (Kasha’s rule) (15), which is currently understood to be a consequence of the increased density of states, which provides a richer landscape of potential conical intersections; indeed, highly excited molecules are speculated to relax through a sequence of successive conical intersections (16, 17). Such pathways are thought to play an important role in explosive molecules (5, 18).

For polyatomic molecules, the potential energy surfaces of the highly excited states that can be accessed by VUV pulses are typically different from those of the ground state. Therefore, the wavepacket experiences a force in the direction of the gradient, leading to fast structural changes in the first few femtoseconds after photon absorption. Along its way, the wavepacket can encounter a conical intersection seam (or a region of strong nonadiabatic coupling), where an efficient transfer of population to another electronic state can occur. The act of “passing through” a conical intersection involves a change in the electronic wavefunction, such that the vibrational wavepacket subsequently evolves on a different potential energy surface. The timescale for the nonadiabatic relaxation depends on coupling strength, the proximity of the strong nonadiabatic coupling region, and the topology of the conical intersection. If the conical intersection lies close to the Franck-Condon region and has a peaked topology (a pronounced funnel character) (12, 19), the relaxation can be on ultrafast (tens of femtoseconds) timescales. Experiments probing such dynamics require ultrafast (∼20-fs) time resolution to capture these coupled electronic-vibrational dynamics. The time-resolved photoelectron spectroscopy technique has been widely used to monitor internal conversion through conical intersections (20–22), mainly for low-lying excited states. The nonadiabatic lifetimes can vary from several tens of femtoseconds, for internal conversion between low excited states of pyrazine (23) or pyridine (24), for a few hundred femtoseconds, which is the typical timescale for the nonradiative relaxation of polyatomic molecules to their ground state (25).

In this work, we implemented ultrafast VUV-driven photoelectron–photon coincidence (PEPICO) imaging spectroscopy to investigate the highly excited neutral states of a model energetic molecule, methyl azide. Ultrafast (10-fs) high harmonic pulses at 8 eV were used to prepare a single highly excited state that is probed via multiphoton ionization by an IR laser field (Fig. 1). We found that when methyl azide is excited in the VUV at 8 eV, the vibrational wavepacket is nonadiabatically accelerated (i.e., accelerated faster than would adiabatically occur), leading to ultrafast internal conversion through a conical intersection to a dark state. This population transfer was complete on a surprisingly fast timescale of 25 fs, which is much faster than the adiabatic vibrational period of 135 fs for the most closely related vibrational mode, the CNN angle bend. These dynamics were directly evidenced by a very rapid increase in vibrational energy in the molecule, followed by a fast decrease in ionization signal induced by rapid population transfer to a dark state.

To uncover the mechanisms behind this ultrafast relaxation, we implemented a set of complementary advanced theoretical calculations to describe the excited states and resulting dynamics. These calculations showed that the initially prepared excited electronic state is mixed with both Rydberg and valence character. This excited electronic state strongly interacted with neighboring lower excited states through nonadiabatic couplings that drove large changes in the electronic and structural character of the molecule on an ultrafast timescale, ∼10 fs. The wavepacket was quickly transferred to a lower excited state along one of the conical intersection seams, which led to a faster relaxation timescale than if the wavepacket had to travel all of the way to the MECI region. In fact, these nonadiabatic couplings were already strong in the Franck-Condon region, leading to a faster timescale for relaxation than is typical for the lowest excited state, when the time required for the wavepacket to reach a strong nonadiabatic region is typically longer (14, 26). These findings are important for understanding the unique strongly coupled non-Born–Oppenheimer molecular dynamics of highly excited states of neutral molecules.

**Experimental Methods**

For our experiments, we used an amplified Ti:Sapphire laser producing pulses at a wavelength of 780 nm with 1.5-mJ energy and 35-fs duration at a repetition rate of 10 kHz. Part of the laser output (200 μJ) was used to generate VUV pump pulses by focusing the beam into a xenon-filled hollow waveguide designed for high harmonic generation, producing odd harmonics of the driving frequency (5ω, 7ω) (27). To eliminate harmonics >7ω, we sent the beam through a 10-cm-long region filled with 3 torr of krypton. The fifth (8.0 eV) and seventh (11.2 eV) harmonics were then focused by using dielectric mirrors coated for 155 nm (Acton Optics), which were found to also reflect a useful amount of 7ω. The other part of the laser was attenuated to 20 μJ and used as a time-delayed probe pulse. The focused probe intensity was 9 × 10^12 W/cm^2, well below the tunneling regime, but with sufficient intensity to drive two-photon processes. Harmonics were generated on the leading edge of the laser pulse, which also experienced self-phase modulation and blue-shifting due to ionization of the gas. Thus, they can shift in energy slightly during the high harmonic generation process (28, 29) and must be independently measured. In this work, the energy of 7ω was determined from the photoelectron spectra of known compounds (acetone and methyl azide), while 5ω was measured on a home-built spectrometer.

To measure electron and ion energies in coincidence, we used an imaging PEPICO spectrometer, shown schematically in Fig. 1, which has been described in detail (30). The detectors could measure both position and time-of-flight of all of the imaged ions and electrons. We recorded each ion–electron pair individually, keeping count rates low (0.1 counts per pulse average) to avoid false coincidences. Position and time-of-flight information allowed us to determine the full 3D momentum vector for each particle.

Optical choppers were placed in the path of each beam so that signal could be alternately collected with either the pump pulse or the probe pulse, or with both pulses present. In the limit of no dark current or signal saturation, the pump–probe signal is the signal with both pulses present minus the signal from each pulse by itself. The ionization potential of methyl azide (9.8 eV) is between harmonics 5ω and 7ω, so that only 7ω produces ions when the probe pulse is blocked (Figs. 1 and 3). When both pulses are present, 5ω creates a neutral excited state that is subsequently ionized by two photons from the probe pulse. An alternative signal pathway, where the IR probe interacts with a cation state prepared by 7ω, was ruled out by repeating the experiment with a thin film of glass (SiO2, Lebow) placed in the pump beam path to remove 7ω and any higher harmonic. The involvement of more than two probe photons was ruled out by again repeating the experiment with a 390-nm probe (instead of 780 nm), with identical results aside from degraded time resolution.

Methyl azide was introduced into the ultrahigh vacuum chamber through an effusive nozzle with a backing pressure of 1 torr, approximately the vapor pressure of methyl azide held at −78 °C by an isopropanol/dry ice slurry (see Methods for synthesis and safe handling procedures of methyl azide).
Computational Methods

Accurate ab initio modeling of the full experiment—excitation, nonadiabatic evolution, photoionization, and subsequent dissociation of the cation—is computationally out of reach for any given level of theory. The strategy here was to combine several different levels of theory that could each tackle a part of the problem reliably, described briefly here and in more detail below. Geometry optimizations and single-point excited-state wavefunction calculations were carried out with the coupled-cluster (CC) methodology at a very high level of accuracy to identify the excited state accessed by the pump pulse. The mixed quantum-classical trajectory guided ab-initio multiple spawning (AIMS) method (31, 32) was then used to follow the dynamics of a wavepacket launched on this state and its relaxation to lower excited states. The photoionization probabilities were computed for all of the excited states involved in the dynamics to determine which excited states could be probed by the second pulse. Finally, the dissociation of the cation was modeled with the Gaussian suite of electronic structure programs (33). The complete basis set–atomic pair natural orbital (CBS-APNO) extrapolation method of Montgomery et al. (34) was used to compute accurate reaction barriers and bond energies. The CBS-APNO calculations have a mean absolute deviation of 0.5 kcal/mol (0.02 eV) for heats of reaction. By combining this complementary set of calculations, a full picture of both the structure and the dynamics for the full experiment could be pieced together.

CC calculations (35, 36) were performed with the PSI4 quantum chemistry program (37) to identify the nature of the excited state. Ground-state CC theory at the singles, doubles, and perturbative triples level CCSD(T) along with the aug-cc-pVTZ basis set (38, 39) was used to optimize an initial geometry. From this structure, equation-of-motion CCSD (EOM-CCSD) (40, 41) excited-state computations were undertaken. As no experimental absorption spectrum of methyl azide was available in the 8-eV energy range accessed by the pump pulse, the appropriate level of theory was chosen by also calculating the closely relatedHN2, whose VUV absorption spectrum is known (3f Appendix). A doubly augmented, triple-zeta basis set (d-aug-cc-pVTZ) was necessary for adequate diffuseness of the basis set to properly describe the spatial extent of the system such that good agreement with experimental data for these highly excited states could be achieved. Such diffuse orbitals have been used with much success (42–44). By using this d-aug-cc-pVTZ basis set, states are accurate to within 0.1 eV.

The nonadiabatic dynamics following the photoexcitation by the pump pulse has been modeled by using the AIMS method, which describes simultaneously the electronic and nuclear dynamics and accounts for the nonadiabatic coupling between electronic states. In AIMS, the time-dependent wavefunction is expressed by a sum of products of a time-dependent nuclear wavefunction and an electronic wavefunction. For each electronic state, the time-dependent nuclear wavefunction is

![Diagram](image)

**Fig. 1.** Femtosecond PEPICO spectroscopy of methyl azide Rydberg-valence states. **A** shows two laser pulses which interact with an effusive molecular beam; the pump pulse prepares an excited state which is then ionized by the probe pulse. Ionized fragments and their coincident electrons are detected with time- and position-sensitive detectors. **B** shows a schematic of the dynamics in the electronic states. The pump pulse excites a wavepacket on a highly excited state (S8). If the second pulse probes the dynamics at time t = 0 (B, i), it ionizes the molecule to both the cation ground and first excited states. For pump–probe delay before 25 fs (B, ii), the wavepacket has time to evolve on the excited-state potential energy surface. Due to the corresponding increase in vibrational energy, the probe pulse can only ionize to the cation ground state. For longer pump–probe delays (B, iii), the wavepacket has completely relaxed to the dark S8 state, due to strong nonadiabatic coupling, and the ion yield drops to zero. **C** shows a schematic representation of the potential energy surfaces of the mixed Rydberg-valence state accessed by the pulse and the dark valence state, as well as the conical intersection seam through which the population transfer occurs. The motion of the initial wavepacket on these surfaces is also shown with a white line, as well as points corresponding to the dynamics described in B. **D** shows excitation orbitals computed with EOM-CCSD for the initially prepared excited state, S8, and the dark state, S4. The initial S8 state has mixed (HOMO-1)–excited character, with each component displaying mixed Rydberg and valence character. The dark S4 state is a (HOMO-1)–excited valence state. Isocontours are 0.03 e/Å2 for the mixed Rydberg–valence orbitals and 0.07 e/Å2 for the pure valence orbitals.
expressed as a linear combination of trajectory basis functions, which are multidimensional frozen Gaussians whose positions and momenta are propagated classically following Hamilton’s equations of motion. When a trajectory basis function enters in a region of strong nonadiabaticity (i.e., a region with a strong coupling between electronic and nuclear motion), a new basis function is spawned on the coupled electronic state, allowing for amplitude transfer. The amplitudes are obtained by solving the time-dependent Schrödinger equation in the basis of the trajectory basis functions. In AIMS, the potential energy surfaces and nonadiabatic couplings are computed on the fly at each time step. Additional information on the AIMS methodology can be found in the following reviews: refs. 32 and 45. For the simulations of methyl azide, we used AIMS as implemented in MOLPRO (46, 47), where the electronic structure and nonadiabatic couplings were computed at the state-average multi-configuration self-consistent field (SA-MCSCF) level (SI Appendix). The dynamics were launched on the excited state accessed by the pump pulse for a swarm of 66 initial conditions sampled from a Wigner distribution of methyl azide’s ground state. The simulations were run for 20 fs with a time step of 5 a.u., unless 80% or more of the population had been transferred to lower excited states.

The photoionization lifetimes—i.e., the time it takes to ionize 63% of the population under illumination by an electric field—were computed for the photoionization of all of the neutral excited states to the cationic ground state. The electronic structure of the neutral excited states and the cation was computed at the time-dependent density functional theory (TDDFT) level with the Coulomb-attenuating method (CAM) functional CAM-B3LYP (48) and the basis set 6–31+G(d) in Gaussian09 (33). The excitation energies and character of the excited states were in good agreement with the other electronic structure methods (SI Appendix). The lifetimes are inversely proportional to the square modulus of the photoionization coupling elements, which involved the computation of the dipole moment between the square module of the photoionization coupling elements, which involved the computation of the dipole moment between the wavefunction of the ionized electron (described as a plane wave) and the Dyson orbitals (49, 50). The Dyson orbitals provide a correlated view of the orbital from which the electron has been ionized and play a crucial role in the interpretation of photoelectron lifetimes and angular distributions.

The Gaussian suite of programs (33) was used for the cation potential energy surface characterization. The geometries of the minima and the transition states were optimized by Hartree–Fock, hybrid density functional (B3LYP (51–53) and CAM-B3LYP (48)), second-order Möller–Plesset perturbation theory (54), and quadratic configuration interaction (55) methods. The spin-restricted vs. unrestricted stability of each structure was tested by using standard methods (56). The CBS-APNO extrapolation method of Montgomery et al. (34) was used to compute accurate energy differences. The CBS-APNO calculations have a mean absolute deviation of 0.5 kcal/mol (0.02 eV) for heats of reaction.

Results
Here, we first discuss our time-resolved experiments on the 8-eV excited state of methyl azide. To help interpret the time-resolved data, we then discuss single-photon ionization experiments conducted with the same total photon energy. For further insight, next, we turn to calculations. Separate calculations were performed to understand which excited state was excited by the pump pulse, how that excited state evolved nonadiabatically, how the probe pulse interacted with the wavepacket, and how the subsequent cation evolved and dissociated. The calculations allowed us to interpret our data to show that the electronic relaxation was driven by strong nonadiabatic coupling and that population transfer occurred along a seam well above the MECI.

VUV Pump–IR Probe of Methyl Azide Excited State Dynamics. We first discuss experiments where we used a VUV pump to prepare excited states of the neutral molecule, which were then probed by using a time-delayed IR pulse to ionize. Fig. 2 shows VUV pump–IR probe measurements using 500 (8.0 eV) pump photons followed by a 2 × 103 IR ionizing beam, which delivered a total of 11.2 eV to the system. The mass spectrum contained two signals: \( m/z = 57 \) amu, which is the methyl azide parent ion (CH3N3+), and \( m/z = 28 \) amu, which is a fragment ion. At our combined photon energy (11.2 eV), the only energetically accessible, stable mass-28 ion was HCNH+. Fig. 2A shows the time dependence of the two channels in the mass spectrum, revealing very fast dynamics. The molecular ion channel (green) appeared at \( t = 0 \) and decayed with a 21 ± 5 fs time constant. The fragment ion channel (red) rose with a 21-fs time constant and decayed with a 27 ± 5 fs time constant. This channel had an overall signal size ~7.5 times larger than the molecular ion channel. Overlaid on each signal is a fit curve shown as dashed black lines. As described in SI Appendix, each curve was fit with its decay time as the only adjustable parameter and was convolved with a Gaussian instrument response; the decay time found for the green curve was used as a fixed rise time for the red curve.

Having measured the ion transients, we next discuss the photoelectron spectra in coincidence with the parent ion. Pump–probe photoelectron spectra in coincidence with the parent ion are shown in Fig. 2B. These spectra show a single peak, starting at 1.01 eV at ~40 fs and shifting linearly to 0.91 eV at ~400 fs. By using the total photon energy (11.2 eV) minus the literature ionization energy (57) (9.8 eV) and the photoelectron kinetic energy (1.0 eV), the cation state reached by the probe must have 0.4 eV internal energy at early pump–probe delays and 0.5 eV at longer times. The lowest excited electronic state of the cation is 1.1 eV above the cation ground state (57), which ruled out electronic excitation. The observed 0.4–0.5 eV internal energy must then be vibrational energy.

Next, turning to the fragment ion, pump–probe photoelectron spectra in coincidence with the fragment ion channel are displayed in Fig. 2C. At early times, these spectra showed a broad feature at ~0.5 eV that corresponded to the vibrationally excited cation ground state and a spike at near-zero energies. The cation first excited electronic state, which has a vertical ionization potential of 11.4 eV and a Franck–Condon envelope that covers 10.5–11.9 eV, was a likely assignment for this spike at near-zero energies. The presence of ionization to both cation states suggested that the excited state has mixed character: Both the HOMO and the HOMO-1 electrons were involved in the excited state.

The fragment photoelectron data, too, showed a drift to lower electron kinetic energies, such that at long times (~75 fs), the spike near zero could not be observed anymore. This quasiballistic wavepacket moved into a region of the neutral excited state where we could not energetically access the cation excited state with our probe, but still had access to the cation ground state.

At first glance, the linear drop over time of the photoelectron kinetic energy would seem to have suggested that our signal decay may have been due to the wavepacket moving outside our spectral range (i.e., the energy gap between excited state and cation potential energy curves increasing to above our photon energy). However, the photoelectron energies were dropping at a rate of 1.2 meV/fs and would have taken 800 fs to drop from 1 eV to below threshold. Since the signal decay was much faster than that (25 fs), we concluded that signal decay was not due to the wavepacket motion on the excited state surface, but was due to some other process, such as internal conversion to a lower electronic state.

Single Photon Ionization of Methyl Azide. In pump–probe photoionization experiments, interpretation of the data relies heavily on knowledge of the final state populated after both pump and
probe pulses. Therefore, we turned to single-photon measurements to directly characterize the cation states. Here, we used $7\omega$ (11.2 eV) to directly ionize the molecule and record the same PEPICO observables studied above. We observed the same mass spectrum (Fig. 3B): $m/z = 57$ amu ($\text{CH}_3\text{N}_3^+$) and $m/z = 28$ amu ($\text{HCNH}^+$).

Photoelectron spectra taken in coincidence with each mass signal are plotted in Fig. 3C. Electrons in coincidence with the mass 57 peak are shown in green with a binding energy of 9.9 eV (electron kinetic energy of 1.3 eV) and agree within 0.1 eV with the known vertical ionization potential of methyl azide (57). Electrons in coincidence with the fragment ion are shown in red and overlap those from the molecular ion but increase in amplitude all of the way to the photon threshold at 11.2 eV. The low binding energy part of this spectrum indicated that vibrationally excited levels of the ground electronic state of the cation have enough energy to cause dissociation, and the high binding energy part of the spectrum indicated that we also accessed the cation excited state. Literature photoelectron spectra show a vertical ionization potential of 11.4 eV for the cation excited state, but with a significant Franck-Condon width such that the low-energy edge of the cation excited state begins at $\sim$10.9 eV (57), easily accessible with our 11.2-eV photons. A breakdown diagram constructed from our data gives the appearance energy for the fragment at 10.2 eV, significantly lower than the prior literature value (10.5 eV) (58). The discrepancy was likely due to difficulties distinguishing counts originating from the effusive jet from those originating from the background.

The kinetic energy of the ion fragment is shown in Fig. 3D. The relatively large amount of translational energy, which peaked at 0.15 eV and extended up to $\sim$0.8 eV, explains the width of the $m/z = 28$ amu peak in the mass spectrum. This dissociation is slightly anisotropic ($\beta = 0.15$; Fig. 3D, Inset). In dissociative ionization, a time constant cannot be directly inferred from the $\beta$-parameter as is done in neutral photodissociation (59); however, we could infer that the dissociation was comparable to or faster than molecular rotations.

As discussed above, the photoelectrons in coincidence with the $m/z = 28$ fragment indicated that the fragment could be formed from either a vibrationally excited cation ground state or from the cation excited state. Fragment translational energy curves were nearly identical in shape for the two cases, with a distribution peaking at 0.15 eV if the cation ground state was initially populated (shown in Fig. 3D) and 0.18 eV if the cation excited state was initially populated (SI Appendix). This suggests that either the two electronic states dissociate very similarly, which is possible but seems unlikely, or that the cation excited state internally converts to the cation ground state and then dissociates.

Comparing the single-photon data (Fig. 3) directly to the pump probe data (Fig. 2), we saw that the photoelectron spectra were qualitatively the same but shifted, and the ion kinetic energy release distributions were identical. These observations indicated that the dissociation events observed in the pump–probe experiment occurred in the cation manifold and that during the pump–probe time delay, the system remained as a bound methyl azide molecule. The ultrafast relaxation of the signal, then, was
not due to the dissociation of the neutral, but rather to an internal conversion process to a lower electronic state.

Ab Initio Calculations of Neutral Excited State Character Populated by the VUV Pump Pulse. To identify the excited state populated by the pump pulse, we turned to high-accuracy CC calculations. The singlet excited states of methyl azide up to ~8 eV include several valence and Rydberg states, including mixed states, which are built largely from the split π pair of the azide moiety (in methyl azide, the in-plane component of the azide π pair becomes a C–N σ bonding orbital, σCN, while the out-of-plane component remains a π nonbonding orbital, πnb). Valence transitions from each of these orbitals to each of the split π* pair (into an in-plane σ* component and an out-of-plane π*) as well as transitions to low-ν Rydberg states converging to each of the two lowest cation states occur in the UV to VUV spectral range. The ground and first excited cation states are well described by a Koopmans’ picture of the removal of an electron from πnb and σCN, respectively (this picture breaks down in higher cation states, where hole mixing becomes important [60]).

The excited state populated at 8 eV was the eighth excited singlet state (labeled 3Aσ* or S8) and was a mixture of two Rydberg states converging to different cation electronic states: πnb (HOMO) → 3p and σCN (HOMO-1) → 3s (Fig. 1D), which converged to the cation ground and excited states, respectively. Each of these wavefunction components also displayed some valence character, with 3s being strongly mixed with σ* and 3p mixed with π*. This qualitative description of the wavefunction appeared at all levels of theory applied, including TD-DFT, MCSCF, and EOM-CC. The optimized geometries for the ground-state neutral, the S8 excited state, and the two lowest cation states are given in Table 1. The only pronounced geometry change found in the excited state was in the CNN bond angle, which had an equilibrium value of 127°, somewhat intermediate between that of the πnb and σCN-hole cation states, due to the approximately equal contribution from each Rydberg state. The adiabatic potential energy minimum was 0.5 eV below the Franck–Condon region, agreeing with the experimentally observed amount of vibrational energy (Fig. 2 B and C compared with Fig. 3C). The mixing of the two Rydberg series led to dipole-allowed photoionization from the excited state to both cation states, as was seen in the pump-probe photoelectron data (Fig. 2C). However, the vibrational mode most closely matching the dominant geometry change had an excited-state vibrational frequency of 247.10 cm⁻¹, corresponding to a period of 135 fs, much slower than the observed dynamics.

The experimental measurements indicated that there was fast relaxation to a dark state for photoionization by our probe pulse. Below, we identify this state as S4 (3Aπ*); here, we describe the electronic character of this state and postpone a discussion of how we identify the state until Dynamical Simulations of the Nonadiabatic Relaxation in the Neutral Excited State. S4 was composed of a single excitation σCN (HOMO-1) → π* and was of valence character (Fig. 1D). Due to its configuration, it had no dipole-allowed ionization to the cationic ground state (πnb-hole), but had a dipole-allowed ionization to the lowest cationic excited state (σCN-hole). This made the S4 state a dark state for our

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Table 1. Important bond lengths and angles at the equilibrium geometries of the electronic states involved in pump-probe signal pathways

<table>
<thead>
<tr>
<th>Geometric feature</th>
<th>S8</th>
<th>S4</th>
<th>D0 (e⁻¹)</th>
<th>D1 (σ⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>r(C–N1)/Å</td>
<td>1.484</td>
<td>1.499</td>
<td>1.477</td>
<td>1.483</td>
</tr>
<tr>
<td>r(N1–N2)/Å</td>
<td>1.250</td>
<td>1.256</td>
<td>1.327</td>
<td>1.207</td>
</tr>
<tr>
<td>r(N2–N3)/Å</td>
<td>1.145</td>
<td>1.139</td>
<td>1.126</td>
<td>1.157</td>
</tr>
<tr>
<td>(\angle\text{CNN}')</td>
<td>113</td>
<td>127</td>
<td>113</td>
<td>151</td>
</tr>
<tr>
<td>(\angle\text{NNN}')</td>
<td>174</td>
<td>178</td>
<td>169</td>
<td>174</td>
</tr>
</tbody>
</table>

S8 was calculated at the CCSD(T)/aug-cc-pVTZ level; S4 was EOM-CCSD/d-aug-cc-pVDZ level; and the radical cations were computed with CCSD(d-aug-cc-pVDZ). For the cation states, the orbital ionization occurred from is indicated in parentheses, with a \(-1\) exponent to indicate an electron has been removed.
probe pulse, as 3.2 eV was not enough to ionize $S_4$ to the cation excited state; nonadiabatic population transfer to $S_4$ explained the decay of the ion signal. Between the $S_1$ and $S_8$ states, there were two Rydberg states ($S_6/4A^+$ and $S_7/5A^+$) composed of excitations from $\pi^{*b}$ to in-plane 3p orbitals and one mixed state ($S_5/2A^+$). The $S_6$ state had the same excitations as $S_5$, but was dominated by the $\pi^{*b}$ (HOMO) $\rightarrow$ 3p component, unlike in $S_5$ where $\pi^{*b}$ (HOMO) $\rightarrow$ 3p and $\sigma_{CN}$ (HOMO-1) $\rightarrow$ 3s were more evenly mixed, with $\sigma_{CN}$ (HOMO-1) $\rightarrow$ 3s slightly dominant.

Dynamical Simulations of the Nonadiabatic Relaxation in the Neutral Excited State. To supply additional evidence that internal conversion is the cause of the experimental signal decay, we turned to the calculation of nonadiabatic coupling vectors (NACs) and nonadiabatic wavepacket propagation. First, we focused on the NACs and gradient difference vectors at the equilibrium geometry of the neutral to assess which states were strongly coupled to the $S_8$ state accessed by the pump pulse. The electronic structure of the nine lowest electronic states and NACs were calculated at the SA9-MCSCF(4–8)/6–311+G(d) level by using an active space containing four electrons and eight orbitals: $\pi$, $\sigma$, $\pi^*$, $\sigma^*$, 3s, and a set of 3p orbitals (SI Appendix) and nine states in the state-averaging procedure (SA9). Large nonadiabatic coupling was found between the $S_8$ and $S_4$ states, which are both composed of similar Rydberg and valence excitations, as well as strong coupling from $S_8$ and $S_6$ to $S_4$, which originates from the $\sigma_{CN} \rightarrow 3s$ part of the $S_4$ and $S_8$ states. Since the $S_8$ state was dark for photoionization to the ground-state cation by the probe pulse, the relaxation to this state explained the experimental drop of ion yield as the pump–probe delay increased. No strong coupling was found between either $S_4$ or $S_6$ to the pure Rydberg state $S_8$. From these results, it seems likely that the $S_8$ excited state may be related to internal conversion to either $S_4$ or $S_6$ or potentially through $S_8$ then to $S_4$. However, no experimental signature was observed, within resolution, that clearly indicated the involvement of $S_6$, which was not a dark state with our probe pulse. The timescale of the nonradiative relaxation, which is closely related the decay of the experimental signal, was determined from dynamical simulations carried out with the AIMS method. At the MCSCF(4–8)/6–311+G(d) level of theory used to compute the NACs above, we encountered severe convergence problems for the computation of these NACs between Rydberg states at geometries other than the equilibrium. To avoid this issue, we used a smaller active space without the 3p orbitals [SA6-CAS(4–5)/6–311+G(d)] and a smaller number of states (six instead of nine) in the state-averaging procedure (SA6), which removed states $S_3$ or $S_7$ entirely from the band and collapsed $S_6$ and $S_8$ into a single state of dominantly $\sigma_{CN} \rightarrow 3s + \sigma^*$ character. This new state was labeled $S_{SAS}$, where the SAS subscript stands for small active space, and was taken to represent the $S_8$ state in the dynamical simulations. The other excited states ($S_{SAS}$ to $S_{SAS}$) exhibited the same character as the $S_1$ to $S_7$ states in EOM-CC or MCSCF with the larger active space. At the ground-state equilibrium geometry, $S_{SAS}$ was found to be strongly coupled to $S_{SAS}$, with very similar NAC vectors to those seen in the larger active space (see SI Appendix for further information). We took this to indicate that the forces acting on the wavepacket, at least in the Franck–Condon region, are determined largely by the $\sigma_{CN} \rightarrow 3s + \sigma^*$ component of the wavefunction and depend little on the (neglected) $\pi^{*b}$ $\rightarrow$ 3p component; we thus proceeded with this reduced active space for the characterization of the conical intersection and the modeling of the nonadiabatic dynamics following the photoexcitation.

We identified a MECI between these two states with a peaked topology, emphasizing the funnel character of the population transfer (Fig. 4C). The geometry at the MECI had a large C–N1–N2 angle (141° rather than 113°) compared with the equilibrium geometry of MeN$_2$ ground state (Fig. 4A). The gradient difference vector followed, among others, the direction of the C–N1–N2 angle opening (Fig. 4B). This geometry variation was expected to occur during the excited state dynamics. The time-dependent mean value of the C–N1–N2 angle obtained from the nonadiabatic dynamical simulations is shown in Fig. 5B.

The dynamical simulations were carried out at the SA6-CAS(4–5)/6–311+G(d) level from the $S_{SAS}$ state ($\sigma_{CN} \rightarrow 3s + \sigma^*$ excitation, which is taken to represent the $S_8$ state of the larger active space) and corresponds to excited state accessed by the pump pulse. Due to the strong NACs already present in the Franck–Condon region, the electronic population transfer to the dark $S_{SAS}$ valence state was observed to occur on an ultrafast timescale much faster than the vibrational motion. In <20 fs, most of the population had been transferred to the dark state, which is in agreement with the measured drop of ion yield occurring on a similar timescale. As soon as the dynamics started, we observed a fast opening of the CNN bond angle up to ~125° (Fig. 5B), near the adiabatic equilibrium geometry of the excited state. During this motion, the vibrational potential energy dropped by ~0.6 eV (Fig. 5A), agreeing with both the (adiabatic) EOM-CC geometry optimization and the experimentally observed amount of vibrational energy (0.5 eV). However, the motion occurred in 10 fs, much faster than the adiabatically expected quarter-period of vibrational motion (135 fs/4 ~ 34 fs). The ultrafast population transfer seen here did not proceed through the MECI itself (Fig. 4C), but rather along the seam (12–14) (Fig. 1C and SI Appendix). The MECI geometry had a CNN bond angle of 141°, but this angle only opened to ~125° before the transfer of population to the dark state was essentially finished. We also calculated a decrease of the N1–N2 bond of 0.06 Å during the dynamics, shown in Fig. 5B. The close agreement between the calculated potential energy decrease and the experimental vibrational energy increase provided strong evidence that the calculation correctly describes where on the potential energy surface the population transfer occurs; in
particular, the MECI was located 1.2 eV below the Franck-Condon region, and population transfer from this geometry would be accompanied by significantly more vibrational energy.

**Ab Initio Calculations of the IR Photoionization of the Neutral Excited State by the Probe Pulse.** The nonadiabatic dynamical simulations showed an ultrafast relaxation to the $S_2$ valence state. However, the experiment reports on the ion yield, so we computed the photoionization lifetimes of the neutral excited states to the cation ground state. The electronic structure was computed at the TD-DFT/CAM-B3LYP 6–31+G(d) level, which led to a good agreement of the character and energetics of the excited states with EOM-CCSD/d-aug-cc-pVDZ results (SI Appendix, Table S2). The photoionization lifetimes require the computation of the Dyson orbitals that represent the orbital from which the electron is ionized (49). Since the cation ground state was well described in the Koopmans’ picture by the removal of the $\pi^*$ electron, the Dyson orbital of an electronic state reflects the excitation(s) from the $\pi^*$ orbital.

At the equilibrium geometry of the ground state, the mixed valence–Rydberg and pure Rydberg states ($S_5$ to $S_9$) had very short photoionization lifetimes (between 2 and 25 fs) due to their diffuse Rydberg orbital (50), which indicated that they should promptly ionize during the probe pulse. On the other hand, the $S_1$ valence state exhibited a very long lifetime (>4 ps) because it had no excitation from the $\pi^*$ orbital, so it would not be significantly ionized by the probe pulse. As the probe pulse also ionizes the wavepacket when it leaves the FC region, we investigated the effect of geometrical changes on the photoionization lifetimes. For the $S_4$ and the pure Rydberg states ($S_6$ and $S_7$), the lifetimes remained similar as we approached the MECI geometry. The behavior was different for the two mixed valence–Rydberg states $S_8$ and $S_9$. The highest state ($S_9$), which is optically accessed by the pulse, lost its valence character, which did not significantly impact its photoionization lifetime. The other mixed valence–Rydberg state ($S_8$) lost its Rydberg character when approaching the MECI geometry, so its ionization lifetime increased by several orders of magnitude, which may explain the lack of clear experimental evidence for the involvement of $S_8$. (More information on the photoionization is available in SI Appendix.)

**Ab Initio Calculations of Cation States and Dissociation After the IR Probe Pulse.** In PEPICO spectroscopy, a measurement does not end until the ion hits its detector, tens of microseconds after the pump and probe pulses. To interpret these asymptotic processes in the cation, we turned to ab initio calculations of the cation ground-state potential energy surface, at the CBS-APNO level. These calculations indicated that only one dissociation pathway was energetically accessible, a two-step, three-body breakup with a barrier of 10.49 eV. In this pathway, a hydrogen atom migrated from the methyl group to the adjacent nitrogen atom as the N1–N2 bond broke; there was a concerted rearrangement and bond breaking to form the methyleneimine cation (CH$_3$ = NH$^+$ m/z 29) and a neutral nitrogen molecule (N$_2$).

Although the dissociative ionization of methyl azide initially formed CH$_3$ = NH$^+$, no mass 29 signal was observed. This indicated that the methyleneimine cation intermediate was formed with an internal energy sufficient to cause further fragmentation on a submicrosecond timescale; the barrier for breaking the CH bond in methyleneimine cation is 1.35 eV (61), providing us with an estimate for the minimum internal energy of the methyleneimine cation intermediate.

Our ab initio calculations showed that the barrier to dissociation on the cation excited state is ~2.0 eV, which was not energetically accessible with our 11.2-eV photons, and the fragmentation must occur after internal conversion to the cation ground state. The result that the dissociation must occur on the cation ground state explained the similarity between ion fragment kinetic energy distributions recorded in coincidence with electrons from cation ground and excited states.

**Discussion**

In this experiment, a high-energy-density molecule, methyl azide, was promoted into a highly excited neutral state, where there was a dense manifold of electronic states strongly coupled to one another. Experimentally, ionization counts from this excited-state decay within a very fast timescale of 25 fs. Our ab initio calculations showed that the absorbing state at 8 eV had both Rydberg–Rydberg and Rydberg–valence mixing, and the wave-packet moved quickly toward a strong nonadiabatic coupling region. Our wavepacket propagation calculations showed that the motion was dominated by a large-amplitude opening of the CNN bond angle, from 113° to 125° (Fig. 5), that approached a conical intersection seam within 10 fs, thus explaining our experimental observation of a 25-fs excited-state decay. This motion was approximately a quarter-cycle of the corresponding excited-state adiabatic harmonic vibrational mode and would take 34 fs if the wavepacket evolved adiabatically; that the motion occurs in 25 fs indicated that the Born–Oppenheimer approximation fails at a quantitative level to describe even the initial dynamics of the system. Interestingly, the electronic population transfer occurred in 25 fs, despite the fact that the seam of conical intersections was outside the Franck–Condon region.

As the wavepacket was accelerated on the initially excited potential energy surface, the experimentally detected mass spectrum shifted from the parent ion (CH$_3$N$_3$H$^+$) to the fragment ion (HNCH$^+$), as a consequence of the extra vibrational energy in the ion. The actual dissociation event occurred well after the probe pulse, but long before the ion finished traveling down the flight tube of the ion detector.

Note that lower excited states of many organic azides, which correspond to the states populated after the nonadiabatic
dynamics observed here, generally have lifetimes of a few hundred femtoseconds (62–64), while the lowest excited state has a lifetime of a few to several hundred picoseconds (65–67). The asymptotic (longtime) products formed from VUV excitation of methyl azide, which occurred after the short-timescale nonadiabatic dynamics discussed here, have been explored in detail by Wodtke and coworkers (68–71), including minor pathways for ion-pair formation (71) and cyclic N3 formation (70). Indeed, very recently, detailed measurements became available of the neutral dissociation products formed after excitation at 157 nm, very close to our pump wavelength (72). The initial dissociation is dominated by N1–N2 bond cleavage to form N2 and CH3N and a C–N1 bond cleavage to form CH3 and cyclic N3. Both channels produce fragments that continue on to secondary dissociation events.

The behavior expected from an adiabatic point of view can be inferred from the EOM-CC calculations. The excited state produced in the Franck–Condon region at 8 eV in methyl azide was an almost equal mixture of two Rydberg–valence components, ϕ(10) (HOMO) → 3p + ϕσ and σCN (HOMO-1) → 3s + ϕσ. The equilibrium structure of this state approximately averaged the two lowest-energy cation structures, giving an equilibrium CNN bond angle of 127° and an N1–N2 bond length of 1.26 Å and indicating that the Rydberg components dominated the adiabatic behavior. Purely adiabatic dynamics, then, can explain the opening of the bond angle, but not the fast rate of the dynamics. These observations emphasize that purely adiabatic arguments are inadequate, and nonadiabatic calculations must be used to model the experiment.

More generally, the fast population transfer seen here may represent a nearly universal response to small molecules excited by VUV pulses in the 7- to 9-eV energy range. The adiabatic Sr state included a mixing between two diabatic Rydberg states (ϕ(10) → 3p and σCN → 3s) and a diabatic valence state (σCN → ϕσ). The excited state here was an almost 50/50 mix of HOMO-excited and (HOMO-1)–excited character, indicating that we were approaching the strong-coupling limit (i.e., the coupling was large compared with the energy difference between basis states). Although apparently coincidental, a near-degeneracy between low-n Rydberg states of different series may actually be quite common. Using typical quantum defects [6 – 1 for a s-wave and 6 – 0.5 for a p-wave (73, 74)], we would expect HOMO → 3p and (HOMO-1) → 3s to be 400 cm−1 apart at the equilibrium bond length and the first ionization potentials are separated by ~1.2 eV, which is a reasonably common situation for organic molecules (75). Considering the number of potential Rydberg–Rydberg interactions that may occur (i.e., near-degenerate pairs other than HOMO → 3p and HOMO-1 → 3s), it seems likely that near-degeneracies between low-n Rydberg states may be ubiquitous in the 7- to 9-eV energy range in organic molecules.

Another important aspect of the electronic mixing seen in methyl azide is that the two components of the wavefunction differ by two electrons, which affects experimental observables such as the photoelectron angular distribution (PAD). The large majority of femtosecond PAD measurements and studied conical intersections between S1 and S2 states (20, 76), which are usually (but not always) HOMO → LUMO and HOMO → (LUMO + 1). The excited-state wavefunction in these cases can be written as |ϕ(1)ϕ(2) + |ϕ(2)ϕ(1) = |ϕ(1)⟩(|ϕ(1)⟩ + |ϕ(2)⟩), where a “−1” superscript indicates a hole in a usually filled orbital and a “+1” superscript indicates an electron in a usually empty orbital. If the wavefunction can be written this way, the two parts (|ϕ(1)⟩ + |ϕ(2)⟩) interfere, and the PAD becomes a sensitive measurement of electronic changes. Antisymmetrization of the electronic wavefunction does not change this result. However, in the case of methyl azide, the wavefunction is of the form |ϕaϕb + |ϕbϕa), sometimes referred to as “nonconjugate” mixing (74), which causes the PADS from ϕa and ϕb to add as probability densities rather than as amplitudes. This makes the total PAD much less sensitive as an indicator of electronic changes, despite the fact that such a system would normally be considered a case of “corresponding” continua (77, 78). In light of the potentially near-ubiquitous nature of electronic Rydberg–Rydberg mixing and the two-electron differences such mixing creates, the additional information available in PEPICO has advantages over PAD measurements.

Conclusions

Highly excited neutral states near 8 eV of a model energetic molecule, methyl azide, were investigated by femtosecond VUV-pump, IR-probe coincidence spectroscopy and modeled in detail with CBS-APNO, EOM-CC, and MCSF calculations. The initially prepared excited state displayed both electronic Rydberg–Rydberg mixing and Rydberg–valence mixing. It underwent rapid, but specific, nonadiabatic relaxation to a lower-lying valence state on a 25-fs timescale, providing direct experimental observations of the nonadiabatic relaxation of highly excited states with femtosecond resolution. Moreover, these phenomena may be reasonably ubiquitous in the 7- to 9-eV energy range for organic molecules.

Methods

Methyl azide was prepared by established procedures (68, 79), slightly modified to increase safety. It should be noted that although methyl azide is an explosive molecule, the large majority of reported accidents have been traced to either hydrazoic acid or heavy metal azide formation (80). Consequently, the procedure here used clean glassware that had never been used with heavy metal salts, and the use of protic solvents was avoided. Sodium azide was dissolved in DMSO and warmed to 80 °C under flowing helium. An equimolar amount of dimethyl sulfate was slowly injected, causing the reaction mixture to bubble. The gas flow was passed over KOH pellets to remove both any moisture and any hydrazoic acid, then through a liquid nitrogen cold finger, where the methyl azide collected as a bright white powder. Mass spectra of the sample indicated that the product was methyl azide (m/z 57) with a small amount of dimethy ether (m/z 46); the dimethy ether could be removed by warming the sample to ~78 °C on a dry ice isopropanol slurry and pumping overnight. Because methyl azide is shock sensitive, it is recommended to avoid using more than ~10 mg at a time.

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