

Large Pt anomaly in the Greenland ice core points to a cataclysm at the onset of Younger Dryas

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One explanation of the abrupt cooling episode known as the Younger Dryas (YD) is a cosmic impact or airburst at the YD boundary (YDB) that triggered cooling and resulted in other calamities, including the disappearance of the Clovis culture and the extinction of many large mammal species. We tested the YDB impact hypothesis by analyzing ice samples from the Greenland Ice Sheet Project 2 (GISP2) ice core across the Bølling-Allerød/YD boundary for major and trace elements. We found a large Pt anomaly at the YDB, not accompanied by a prominent Ir anomaly, with the Pt/Ir ratios at the Pt peak exceeding those in known terrestrial and extraterrestrial materials. Whereas the highly fractionated Pt/Ir ratio rules out mantle or chondritic sources of the Pt anomaly, it does not allow positive identification of the source. Circumstantial evidence such as very high, superchondritic Pt/Al ratios associated with the Pt anomaly and its timing, different from other major events recorded on the GISP2 ice core such as well-understood sulfate spikes caused by volcanic activity and the ammonium and nitrate spike due to the biomass destruction, hints for an extraterrestrial source of Pt. Such a source could have been a highly differentiated object like an Ir-poor iron meteorite that is unlikely to result in an airburst or trigger wide wildfires proposed by the YDB impact hypothesis.

meteorite impact | climate change | ICP-MS analysis | PGE

The Younger Dryas (YD), a millennium-long cooling period amid postglacial warming well documented in the Greenland ice cores (e.g., refs. 1, 2), is thought to result from an abrupt change in atmospheric and oceanic circulation (3). Whether such a change was caused by a catastrophic event or it is an integral, although still poorly understood, feature of the deglaciation process remains unclear (4).

Among testable catastrophic hypotheses, the most popular, attractive, and long-lasting idea of a sudden discharge of fresh water from the proglacial Lake Agassiz into the Arctic Ocean (5–7) eventually was found inconsistent with geomorphological and chronological data (4, 8). The long-term effect of the proposed “volcanic winter” in the northern hemisphere induced by the catastrophic eruption of the Laacher See volcano 12,916 calendar years before 1950 (cal BP) (9) is not clear as the Laacher See tephra, found in many European lacustrine deposits, is absent in the Greenland ice cores (10). The impact hypothesis (11), once declared dead (12, 13), recently gained new support from the discovery of siliceous scoria-like objects (SLOs) with global distribution, which provide strong evidence for processing at high temperatures and pressures consistent with a cosmic impact (14).

The ever-controversial impact hypothesis was initially invoked to explain the disappearance of the Clovis culture and the extinction of many mammal species, including mammoths, by a cometary airburst that resulted in massive wildfires and, ultimately, the YD cooling. A C-rich layer exposed at many sites in North America and Europe at or near the YD boundary (YDB layer), which is enriched in magnetic grains with Ir, magnetic microspherules, charcoal, soot, carbon spherules, glass-like carbon with nanodiamonds, and fullerenes with extraterrestrial He (11), has been interpreted as documenting this event.

Ammonium and nitrate spikes at the onset of the YD in Greenland Ice Core Project (GRIP) and Greenland Ice Sheet Project 2 (GISP2) ice cores, perhaps resulting from biomass burning, are taken as further support for the impact hypothesis. Subsequent studies (13, 15, 16) questioned the origins of many impact markers cited by ref. 11, but the discovery of the SLOs alongside abundant, compositionally similar microspherules in three YDB sites in North America and Asia is difficult to explain by anything other than a cosmic impact. In its latest incarnation, the impact hypothesis calls for three or more epicenters of an impact or airburst (14). However, the invoked markers have never been supported by a clear geochemical impact signature such as a sharp increase in Ir or other platinum group element (PGE) concentrations at the YDB.

We have tested the YDB impact hypothesis by measuring trace and major element concentrations in ice samples from the GISP2 ice core across the Bølling-Allerød/YD boundary (depth of 1,709–1,720 m, 12,279–13,064 cal BP) with a spatial resolution of ~12.5 cm, corresponding to a time resolution of 2.5–4.6 y (17). The elemental concentrations in melted ice samples were measured by inductively coupled plasma mass spectrometry (ICP-MS) that is known to have possible interferences of LuO and HfO peaks with Ir and Pt peaks. This issue was resolved by measuring Lu and Hf oxide formation in well-calibrated standards (details in *Materials and Methods*).

The major finding of this study (Fig. 1 and Table S1) is the lack of a striking Ir anomaly in the analyzed ice samples. Instead, we found a large Pt anomaly in the middle of the Bølling-Allerød–YD transition that is contemporaneous with a sharp drop in the $\delta^{18}\text{O}$ values (Fig. 1B and Fig. S1). The Pt peak (Fig. 1) is unlikely to be a mass-spectrometry artifact because (a) it is smoothly defined by seven ice samples; (b) the HfO interferences ($^{178}\text{Hf}^{16}\text{O}$, $^{179}\text{Hf}^{16}\text{O}$, $^{180}\text{Hf}^{16}\text{O}$) on $^{194,195,196}\text{Pt}$ isotopes, respectively, were carefully assessed; (c) the $^{178,179,180}\text{Hf}$ signals in sample 63 with the highest Pt concentration are at least a factor of 10 lower than those of $^{194,195,196}\text{Pt}$; and (d) there is no linear correlation between Hf concentrations and Pt concentrations in the ice samples.

The Pt concentrations gradually rise by at least 100-fold over ~14 y and drop back during the subsequent ~7 y. The decay of the Pt signal is consistent with the ~5-y lifetime of dust in the stratosphere. The observed gradual ingrowth of the Pt concentration in ice over ~14 y may suggest multiple injections of Pt-rich dust into the stratosphere that are expected to result in a global Pt anomaly.

The Pt anomaly is accompanied by extremely high Pt/Ir and Pt/Al ratios (Fig. 2), indicative of a highly unusual source of Pt in the ice. Such a source is unlikely to be laboratory contamination

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composition derived from a highly fractionated portion of a proto-planetary core.

Materials and Methods

Sample Preparation. The 11 ice core pieces (~2 cm × 3 cm × 1 m each) provided by the National Ice Core Laboratory were processed in our clean laboratory at room temperature. The ice “sticks” packed in plastic sleeves arrived partially broken into smaller pieces. Before sampling, each stick was carefully aligned in the original sleeve and then split into eight samples, each ~12.5 cm long, using a metal bar wrapped first in aluminum foil and then in plastic wrap. Subsequently, each sample was extracted from the sleeve, carefully washed with ultrapure deionized (DI) water to remove possible surface contamination, and placed in a tall 300-mL polypropylene container, precleaned in 1:1 HCl and then 1:1 HNO₃ for several days. Small ice fragments and water left in a sleeve were composited and used to develop and test analytical protocols. “Composites” from several sleeves were combined to get total weight comparable to that of the real ice samples. In total, 88 ice samples (samples 1–88 in Table S1), ranging from 38.42 g to 68.90 g, and 5 composite samples (samples 89–93 in Table S1) were collected and processed. The empty and filled polypropylene containers were weighed with a precision of ±0.01 g.

Ice samples were melted at room temperature in closed containers and the water was slowly evaporated at 80–90 °C on a hot plate down to 1–2 mL. The remaining samples were carefully transferred into precleaned (boiled in 1:1 HCl and then 1:1 HNO₃) 6-mL, perfluoroalkoxy (PFA) teflon beakers. Each polypropylene container was then washed at least twice with 1–2 mL ultrapure DI water, and the washes were collected and added to the corresponding PFA beakers.

Sample Dissolution. We attempted to measure the total Ir and Pt concentrations in ice, including all solids cemented in ice. Consequently, we applied a multiple-step dissolution method as documented below.

The 93 samples along with 2 procedure blanks were (i) dried down on a hot plate, (ii) sealed with 1 mL of a 1:1 mixture of concentrated HF and concentrated HNO₃ and heated at 150 °C in an oven for 1 wk, (iii) dried down again, (iv) redissolved in 0.4 mL of aqua regia (one part concentrated HNO₃ and three parts concentrated HCl), and (v) finally diluted with 1 mL of ultrapure DI water to yield analytical solutions of ~1.45 g each. Both empty and filled PFA beakers were weighed with a precision of ±0.0001 g.

The HF-HNO₃ step is aimed at breaking down the silicates, and the aqua regia step is aimed at transferring all PGEs into solution because it is well known that Ir and Pt are not stable in diluted HNO₃ (e.g., ref. 27). The latter has been confirmed by our tests using the composite samples. Our approach differs from previous studies such as refs. 28–30, in which ice samples were melted, preconcentrated by subboiling point evaporation, and added concentrated HNO₃ to form 1% HNO₃ analytical solutions for ICP-MS measurements. Such a treatment is unable to transfer all Ir and Pt into solution. Our approach is also different from that of ref. 31, which measured PGE concentrations only in particles in the size range 0.45–20 μm. This explains

why our measurements yield much higher Ir and Pt concentrations in ice compared with those in refs. 28 and 31.

ICP-MS Measurements. All ice samples, in diluted aqua regia matrix, were analyzed in two analytical sessions, using a GV Instruments Platform ICP-MS, first for Ir, Pt, Lu, and Hf with an Apex inlet system to increase sensitivity and reduce oxide interferences and then for major elements with a normal glass spray chamber. For Ir-Pt measurement, peaks of ¹⁷⁵Lu, ¹⁷⁶, ¹⁷⁷, ¹⁷⁸, ¹⁷⁹, ¹⁸⁰Hf, ¹⁹¹, ¹⁹³Ir, and ¹⁹⁴, ¹⁹⁵, ¹⁹⁶, ¹⁹⁸Pt were monitored. Because of the special memory effects of Ir and Pt, diluted aqua regia (2.5 mL concentrated HCl + 7.5 mL concentrated HNO₃ + 90 mL H₂O) was used as a wash solution. Instrumental blank was also measured for such a solution. Oxide formation rates of LuO/Lu and HfO/Hf were determined for a 10 parts per billion (ppb) Lu-Hf standard solution prepared from HPS 1,000-ppm single-element standards before and after analytical sessions. During all measurements, LuO/Lu was <0.01 and HfO/Hf <0.05. Ratios of LuO/Lu (<0.01) and HfO/Hf (<0.05), measured in a Lu-Hf solution together with ice samples, were used to correct for oxide interferences (¹⁷⁵Lu¹⁶O, ¹⁷⁷Hf¹⁶O, ¹⁷⁸Hf¹⁶O, ¹⁷⁹Hf¹⁶O, ¹⁸⁰Hf¹⁶O) on ¹⁹¹, ¹⁹³Ir and ¹⁹⁴, ¹⁹⁵, ¹⁹⁶Pt, respectively. The oxide corrections were typically <5 atomic %. After oxide interference correction, the measured Ir and Pt isotopic compositions of ice samples are within 3% of those measured on the Ir-Pt standard solution, showing that our Ir-Pt measurements were not affected by LuO and HfO interferences.

During our Ir-Pt measurements, the ICP-MS was tuned with a 10-ppb Ir solution, and this solution was analyzed every five samples to monitor the instrumental sensitivity drift, which was less than 10% during any of our analytical sessions. A 10-ppb Ir-Pt standard solution, prepared from HPS 1,000-ppm single-element standards, was analyzed at the end of an analytical session to convert measured signal into concentrations. This is specially designed because of the long memory effects of Ir. For Ir, concentrations obtained using ¹⁹¹Ir and ¹⁹³Ir (after oxide corrections) are within 3%, and an average is used. For Pt, although peak ¹⁹⁸Pt is free of oxide interference, it suffers isobaric interference from ¹⁹⁸Hg. Consequently, we used only ¹⁹⁴, ¹⁹⁵, ¹⁹⁶Pt peaks (after oxide corrections). Concentrations obtained using ¹⁹⁴, ¹⁹⁵, ¹⁹⁶Pt peaks are within 3% of each other, and an average is used. All ice samples were analyzed at least twice in a random order, with the average values reported in Table S1.

For major element measurements (Al), a Basalt, Hawaiian Volcanic Observatory (BHVO)-1 standard solution was measured every five to six samples to monitor the instrumental sensitivity drift, which was less than 10% during any of our analytical sessions. This solution was also used for calibration. All ice samples were analyzed at least twice in random order, with the average values reported in Table S1.

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