

Impact of fossil fuel emissions on atmospheric radiocarbon and various applications of radiocarbon over this century

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Radiocarbon analyses are commonly used in a broad range of fields, including earth science, archaeology, forgery detection, isotope forensics, and physiology. Many applications are sensitive to the radiocarbon (¹⁴C) content of atmospheric CO₂, which has varied since 1890 as a result of nuclear weapons testing, fossil fuel emissions, and CO₂ cycling between atmospheric, oceanic, and terrestrial carbon reservoirs. Over this century, the ratio ¹⁴C/C in atmospheric CO₂ ($\Delta^{14}\text{CO}_2$) will be determined by the amount of fossil fuel combustion, which decreases $\Delta^{14}\text{CO}_2$ because fossil fuels have lost all ¹⁴C from radioactive decay. Simulations of $\Delta^{14}\text{CO}_2$ using the emission scenarios from the Intergovernmental Panel on Climate Change Fifth Assessment Report, the Representative Concentration Pathways, indicate that ambitious emission reductions could sustain $\Delta^{14}\text{CO}_2$ near the preindustrial level of 0‰ through 2100, whereas “business-as-usual” emissions will reduce $\Delta^{14}\text{CO}_2$ to –250‰, equivalent to the depletion expected from over 2,000 y of radioactive decay. Given current emissions trends, fossil fuel emission-driven artificial “aging” of the atmosphere is likely to occur much faster and with a larger magnitude than previously expected. This finding has strong and as yet unrecognized implications for many applications of radiocarbon in various fields, and it implies that radiocarbon dating may no longer provide definitive ages for samples up to 2,000 y old.

fossil fuel emissions | radiocarbon | atmospheric CO₂ | ¹⁴C dating | isotope forensics

Radiocarbon is produced naturally in the atmosphere and decays with a half-life of $5,700 \pm 30$ y (1–3). Fossil fuels, which are millions of years old, are therefore devoid of ¹⁴C, and their combustion adds only the stable isotopes ¹²C and ¹³C to the atmosphere as CO₂. First observed by Hans Suess in 1955 using tree ring records of atmospheric composition (4), the dilution of ¹⁴CO₂ by fossil carbon provided one of the first indications that human activities were strongly affecting the global carbon cycle. The apparent “aging” of the atmosphere—i.e., the decreasing trend in the ratio ¹⁴C/C of CO₂ (reported as $\Delta^{14}\text{CO}_2$) (5)—was interrupted in the 1950s when nuclear weapons testing produced an immense amount of “bomb” ¹⁴C that approximately doubled the ¹⁴C content of the atmosphere. Direct atmospheric observations began in the 1950s, capturing the rapid rise of $\Delta^{14}\text{CO}_2$ and its subsequent quasi-exponential decay as the bomb ¹⁴C mixed into oceanic and biospheric reservoirs (6–9) (Fig. 1).

Now that several decades have passed since the Partial Nuclear Test Ban Treaty and the peak in atmospheric $\Delta^{14}\text{CO}_2$, fossil fuel emissions are once again the main influence on the long-term trend in $\Delta^{14}\text{CO}_2$ (7, 10). The growth or decline of fossil fuel emissions over the coming century determines to what extent $\Delta^{14}\text{CO}_2$ will be diluted further by fossil carbon. Also important is how atmospheric $\Delta^{14}\text{CO}_2$ dilution is moderated by natural exchanges of CO₂ with the ocean and the terrestrial biosphere. Future dynamics of carbon and ¹⁴C are simulated here using the Representative Concentration Pathways (RCPs) developed for the Intergovernmental Panel on Climate Change

(IPCC) Fifth Assessment Report (11, 12), and a simple carbon cycle model with parameters constrained by 20th-century atmospheric and oceanic $\Delta^{14}\text{C}$ and CO₂ observations (10, 13) (*SI Text*).

The simple carbon cycle model includes a one-dimensional box diffusion model of the ocean and represents the atmosphere and the biosphere as one-box carbon reservoirs (10, 13, 14). Exchanges of carbon and ¹⁴C are governed by a small number of model parameters (*Table S1*). Multiple simulations were run using various parameter sets selected by their representation of $\Delta^{14}\text{C}$ and inventories of CO₂ and bomb ¹⁴C (15–17). Atmospheric CO₂ concentration and fossil fuel and land use fluxes were prescribed by the RCPs, which include historical data through 2005. To match the prescribed atmospheric CO₂ concentration, the residual of carbon emissions and atmospheric and oceanic accumulation was added to the biospheric reservoir (single deconvolution). Atmospheric $\Delta^{14}\text{CO}_2$ was prescribed by observations until 2005, then predicted by model fluxes from 2005 to 2100 (*SI Text*).

Results

From its present value of ~ 20 ‰ (18), which signifies a 2‰ enrichment in ¹⁴C/C of CO₂ above preindustrial levels, $\Delta^{14}\text{CO}_2$ is certain to cross below the preindustrial level of 0‰ by 2030, but potentially as soon as 2019 (Fig. 1, *Table S2*). After 2030, simulated $\Delta^{14}\text{CO}_2$ trends diverge according to the continued growth, slowing, or reversal of fossil fuel CO₂ emissions in the RCP scenarios. Distinct patterns are simulated for different RCPs despite the range of model parameters used, indicating the

Significance

A wide array of scientific disciplines and industries use radiocarbon analyses; for example, it is used in dating of archaeological specimens and in forensic identification of human and wildlife tissues, including traded ivory. Over the next century, fossil fuel emissions will produce a large amount of CO₂ with no ¹⁴C because fossil fuels have lost all ¹⁴C over millions of years of radioactive decay. Atmospheric CO₂, and therefore newly produced organic material, will appear as though it has “aged,” or lost ¹⁴C by decay. By 2050, fresh organic material could have the same ¹⁴C/C ratio as samples from 1050, and thus be indistinguishable by radiocarbon dating. Some current applications for ¹⁴C may cease to be viable, and other applications will be strongly affected.

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